

Selected Ions and Charges

POSITIVE IONS

Aluminum	Al ³⁺	Silver	Ag ¹⁺	Bicarbonate	HCO ₃ ¹⁻
Ammonium	NH ₄ ¹⁺	Sodium	Na ¹⁺	Iodate	IO ₃ ¹⁻
Barium	Ba ²⁺	Strontium	Sr ²⁺	Iodide	I ¹⁻
Bismuth (III)	Bi ³⁺	Thallium (I)	Tl ¹⁺	Nitrate	NO ₃ ¹⁻
Cerium (III)	Ce ³⁺	Thallium (III)	Tl ³⁺	Nitride	N ³⁻
Cerium (IV)	Ce ⁴⁺	Tin (II)	Sn ²⁺	Nitrite	NO ₂ ¹⁻
Cadmium	Cd ²⁺	Tin (IV)	Sn ⁴⁺	Oxalate	C ₂ O ₄ ²⁻
Calcium	Ca ²⁺	Titanium	Ti ⁴⁺	Oxide	O ²⁻
Cesium	Cs ¹⁺	Zinc	Zn ²⁺	Perchlorate	ClO ₄ ¹⁻
Chromium (III)	Cr ³⁺	Zirconium	Zr ⁴⁺	Permanganate	MnO ₄ ¹⁻
Cobalt (II)	Co ²⁺			Phosphate	PO ₄ ³⁻
Copper (I)	Cu ¹⁺			Phosphite	PO ₃ ³⁻
Copper (II)	Cu ²⁺			Phosphide	P ³⁻
Gallium (III)	Ga ³⁺	Acetate	C ₂ H ₃ O ₂ ¹⁻	Pyrophosphate	P ₂ O ₇ ⁴⁻
Germanium (IV)	Ge ⁴⁺	Arsenate	AsO ₄ ³⁻	Selenate	SeO ₄ ²⁻
Gold (I)	Au ¹⁺	Bromate	BrO ₃ ¹⁻	Selenide	Se ²⁻
Gold (III)	Au ³⁺	Bromide	Br ¹⁻	Silicate	SiO ₃ ²⁻
Hydrogen	H ¹⁺	Carbide	C ⁴⁻	Sulfate	SO ₄ ²⁻
Iron (II)	Fe ²⁺	Carbonate	CO ₃ ²⁻	Sulfide	S ²⁻
Iron (III)	Fe ³⁺	Chlorate	ClO ₃ ¹⁻	Sulfite	SO ₃ ²⁻
Lead (II)	Pb ²⁺	Chloride	Cl ¹⁻	Tartrate	C ₄ H ₄ O ₆ ²⁻
Lead (IV)	Pb ⁴⁺	Chlorite	ClO ₂ ¹⁻	Telluride	Te ²⁻
Lithium	Li ¹⁺	Chromate	CrO ₄ ²⁻	Thiocyanate	SCN ¹⁻
Magnesium	Mg ²⁺	Citrate	C ₆ H ₅ O ₇ ³⁻	Tungstate	WO ₄ ²⁻
Manganese (II)	Mn ²⁺	Cyanide	CN ¹⁻		
Mercury (I)	Hg ₂ ²⁺	Dichromate	Cr ₂ O ₇ ²⁻		
Mercury (II)	Hg ²⁺	Fluoride	F ¹⁻		
Nickel (II)	Ni ²⁺	Formate	CHO ₂ ¹⁻		
Potassium	K ¹⁺	Hexafluorosilicate	SiF ₆ ²⁻		
Rubidium	Rb ¹⁺	Hydride	H ¹⁻		
Silicon (IV)	Si ⁴⁺	Hydroxide	OH ¹⁻		
		Hypochlorite	ClO ¹⁻		

NEGATIVE IONS